



Chapter 1

Properties of Fluids

- Definition of fluid
- Properties of fluids
- Calculations

Definition of Fluid

All matter can be divided into 2 major classes

- 1) Solid
- 2) Fluids

The difference between a solid and fluid can be defined by

- 1) The stress-strain relationship
- 2) The elasticity

The Stress-Strain Relationship

Solid

- requires external forces to cause it to deform.

Fluid

- will deform without the application of external forces.
- will take on the shape of the container in which they are held.
 - Ex. water, oil, gas etc.

The Elasticity Relationship

Solid

- When external forces stress and deform solids, the **solid will regain their original shape** when these external forces are removed [**Elasticity**].

Fluid

- Fluid will continue to change shape in time** even after the removal of the external forces causing the deformation.



Ideal Fluid

Ideal fluid is a substance that is unable to resist internal shear and tensile forces.



Liquid and Gases

Fluids can be classified into 2 forms of matter

- 1) Liquids
- 2) Gases

The difference between a liquid and gas can be defined by

- 1) The compressibility
- 2) The continuity

Compressibility

:to compress something into a small space.

Liquids

- Liquids are considered to be **incompressible**.
- Liquids only change in volume even when subjected to very high pressure.

Gases

- Gases are very **compressible**.
- Their volume can increase/decrease when subjected to slight variation in pressure.



Continuity

: is the state of being continuous.

Liquids

- When a liquid is held in a container, its **entire mass will arrange itself** so as to be in contact with the bottom and the sides of that container, and **a well-defined surface of the liquid will form.**

Gases

- A **gas held in a closed container will not form a well-defined surface** and will tend to fill the entire container.

Properties of Fluids

Density (ρ)

Density is the mass of the fluid per unit volume

$$\rho = \frac{M}{V}$$

ρ = Density of fluid, kg/m^3

M = Mass of fluid, kg

V = Unit volume, m^3

The density of water at 4°C = $1,000 \text{ kg/m}^3$

Properties of Fluids

Specific Weight/Unit Weight (γ)

The specific weight is the weight of the fluid per unit volume

$$\gamma = \frac{W}{V}$$

γ = Specific weight, N/m³

W = Weight of fluid, N ($W=mg$)

V = Unit volume, m³

The specific weight of water at 4°C = 9.81 kN/m³

Properties of Fluids

Specific Gravity/Relative Gravity (S)

The specific gravity is the ratio of the density or specific weight of the fluid to the density or specific weight of water, at a temperature of 4°C

$$S = \frac{\rho}{\rho_w} = \frac{\gamma}{\gamma_w} \Rightarrow \text{Dimensionless}$$

Properties of Fluids

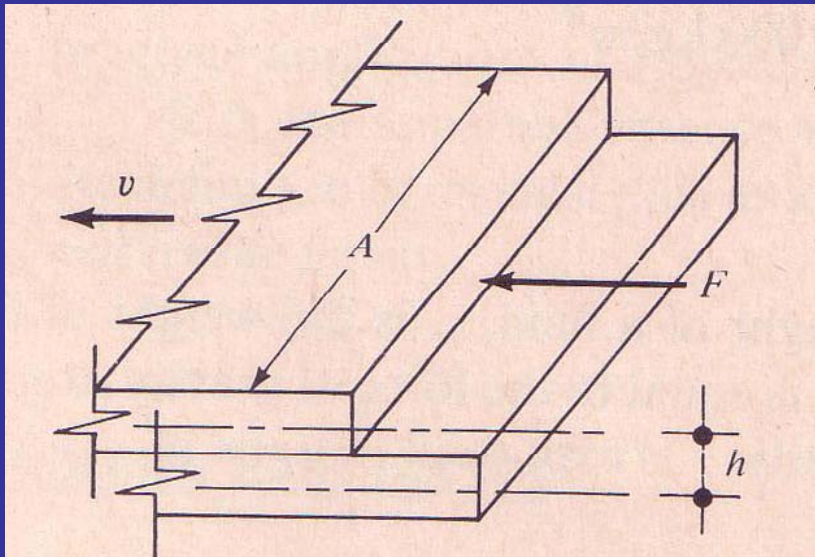
Dynamic Viscosity (μ)

Viscosity is the measure of a fluid's resistance to internal shear stresses.

$$\mu = \frac{\tau h}{v}$$

- μ = Dynamic viscosity, N.s/m² or Pa.s
- τ = Internal shear stress, N ($\tau = F/A$)
- v = Velocity, m/s

Properties of Fluids



A thin layer of fluid

Let

- A = the horizontal area of each layer
- h = the vertical distance between their centerlines
- F = internal shear force

The top layer is acted upon by F

The top layer will move with a velocity, v relative to the bottom layer

Properties of Fluids

Ideal Fluid :

- is unable to resist F
- the relative velocity would remain constant, even F is removed

Real Fluid :

- is able to resist F due to
 - 1) internal molecular activity
 - 2) friction between the layers

Properties of Fluids

$$F = \frac{\mu v A}{h} \quad \tau = F/A \quad \Rightarrow \quad \mu = \frac{\tau h}{v} \quad \Rightarrow \quad \tau = \frac{\mu v}{h}$$

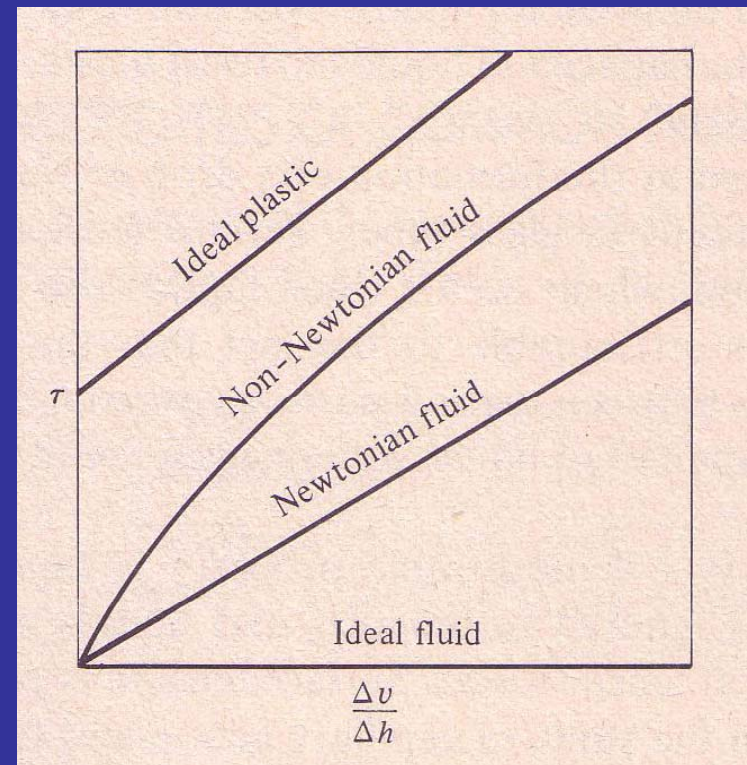
$$\text{If } h \rightarrow 0 \quad \tau = \mu \frac{\Delta v}{\Delta h}$$



Instantaneous rate of deformation of fluid



Newton's Law of Viscosity



Properties of Fluids

Kinematic Viscosity (ν)

The kinematic viscosity of a fluid is the ratio of its dynamic viscosity to its density.

$$\nu = \frac{\mu}{\rho}$$

- ν = kinematic viscosity, m^2/s
 μ = Dynamic viscosity, $\text{N}\cdot\text{s}/\text{m}^2$ or $\text{Pa}\cdot\text{s}$
 ρ = Density of fluid, kg/m^3



Properties of Fluids

Surface Tension (σ)

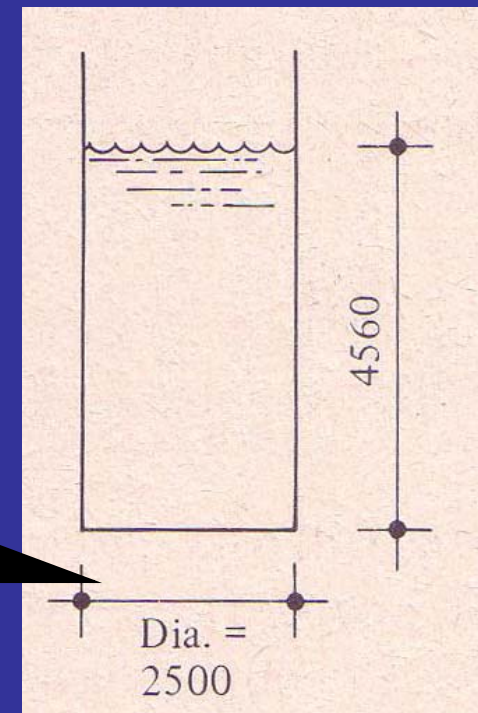
The capacity of liquids to resist tensile stresses at their surface is called surface tensile.

Example 1

Determine the total mass and weight of the fluid in the container in the figure if the fluid is

a) water

b) mercury, with $S=13.6$



$$M_w = 22,400 \text{ kg}, W_w = 220 \text{ kN}$$

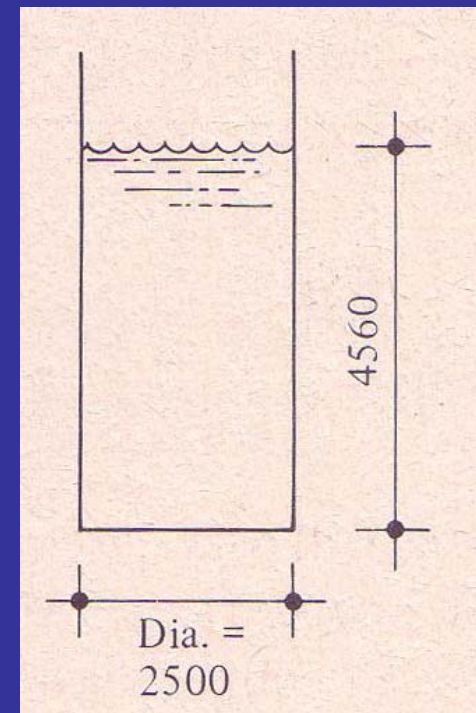
$$M_m = 304,600 \text{ kg}, W_w = 2,990 \text{ kN}$$

Example 2

If the total weight of the liquid in the figure is 505 kN, determine the height of liquid if it is

a) water

b) oil, with $s=0.85$



$$h=10.5 \text{ m}$$

$$h=12.3 \text{ m}$$

Example 3

Determine the dynamic viscosity of an oil with a kinematic viscosity of $0.352 \times 10^{-5} \text{ m}^2/\text{s}$ and a relative density of 0.88.

$$\mu = 310 \times 10^{-5} \text{ Pa}\cdot\text{s}$$

Example 4

A fluid has a specific weight of 9.345 kN/m^3 and a dynamic viscosity of $3.31 \times 10^{-2} \text{ Pa}\cdot\text{s}$. Determine its relative and mass density and its kinematic viscosity.

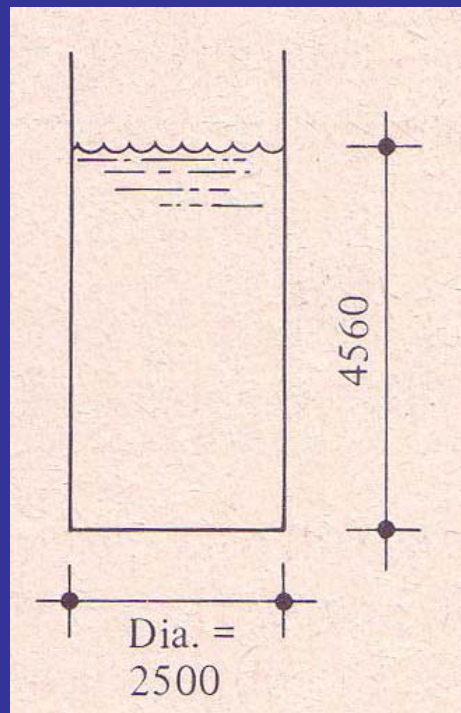
$$S=0.95$$

$$\rho=950 \text{ kg/m}^3$$

$$\nu=3.48 \times 10^{-5} \text{ m}^2/\text{s}$$

Example 5

The fluid of the container in the figure has a total weight of 319 kN and a dynamic viscosity of 91×10^{-5} Pa.s. Determine its relative and mass density and kinematic viscosity.



$$S = 1.45$$

$$\rho = 1,450 \text{ kg/m}^3$$

$$\nu = 6.3 \times 10^{-7} \text{ m}^2/\text{s}$$