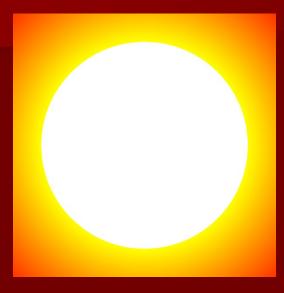
Radiation, Conduction, Convection

An Introduction to Heat Transfer

- I. Heat
 - A. Heat is the transfer of <u>thermal</u> <u>energy</u> from one object to another.
 - B. The amount of heat within an object is measured as <u>temperature</u>, usually in units of <u>Fahrenheit</u> or <u>Celsius</u>.
 - C. Heat moves in one of three ways: <u>radiation</u>, <u>conduction</u>, or <u>convection</u>.



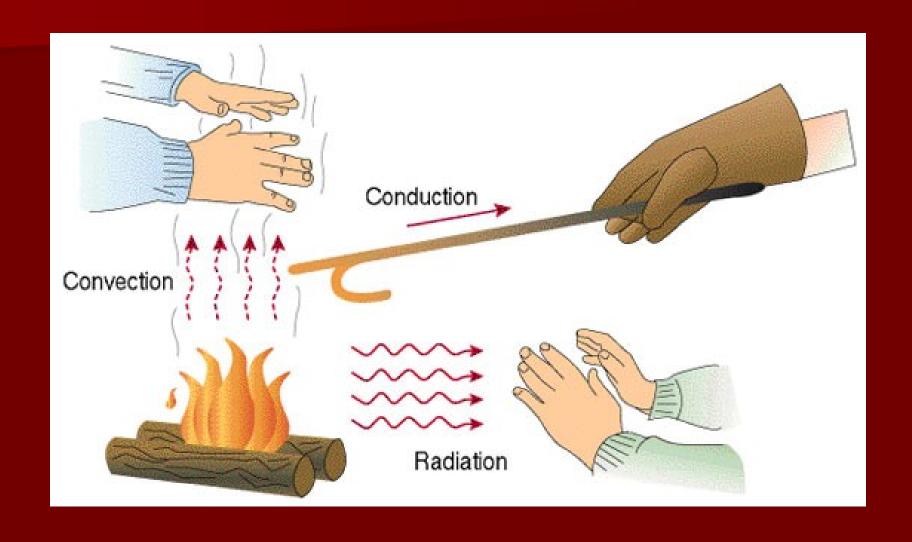




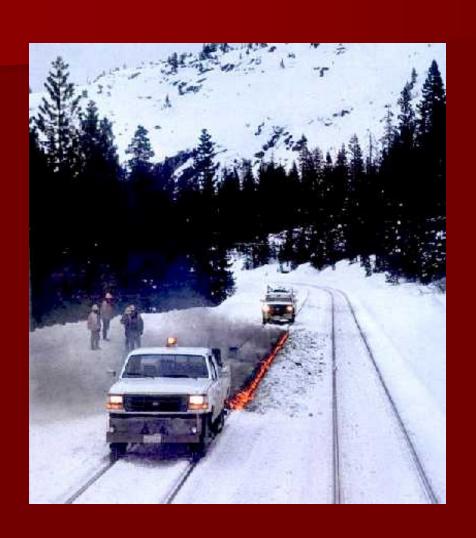


II. Heat Transfer

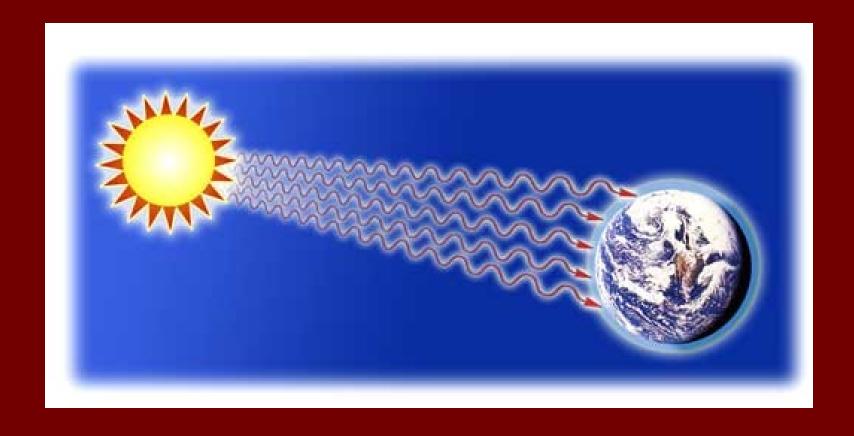
- A. <u>Heat transfer</u> occurs when thermal energy moves from an object with a higher temperature to an object with a lower temperature.
- B. Heat will continue to move until both objects reach *thermal equilibrium*.
- C. The amount of energy in the transfer is measured as *joules*, *BTU's*, or *calories*.



- III. How Heat Transfer Works
 - A. As the temperature of an object increases, the particles inside move faster and begin to move away from each other.
 - B. As the particles move away, the object will *expand* in size.
 - C. As thermal energy leaves the object, the particles move closer together and the object will *contract*.

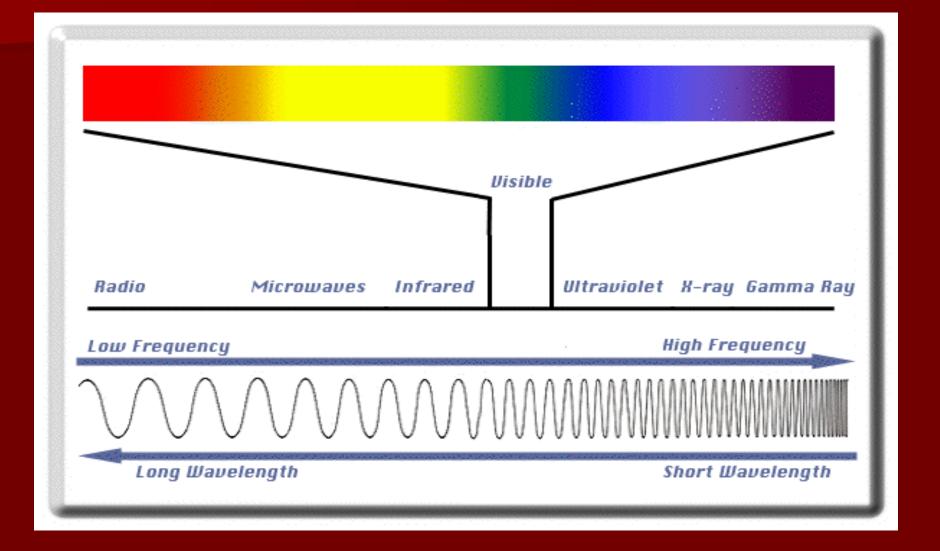




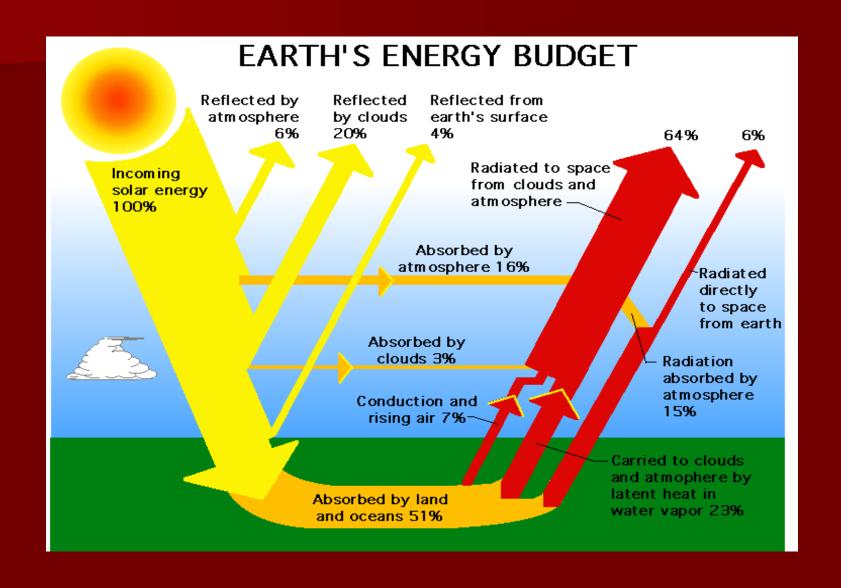


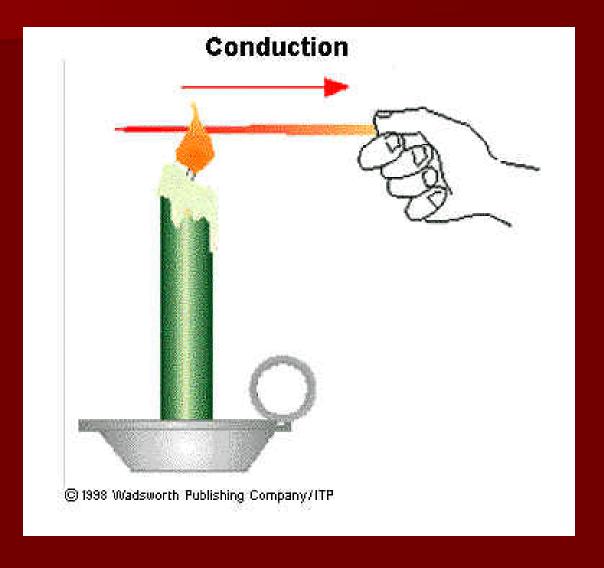
IV. Radiation

- A. <u>Radiation</u> is a form of heat transfer that occurs when <u>electromagnetic</u> waves move through space.
- B. Types of electromagnetic waves:
 - 1. radio waves
 - 2. microwaves
 - 3. x-rays
 - 4. light



- C. Light is a type of electromagnetic wave that carries thermal energy.
 - 1. light from the sun travels 93 million miles to the earth.
 - 2. as light waves enter Earth's atmosphere, they strike particles in the air and ground and <u>disperse</u> their thermal energy.
 - 3. this process warms the air and ground and provides most of the heat for the planet.





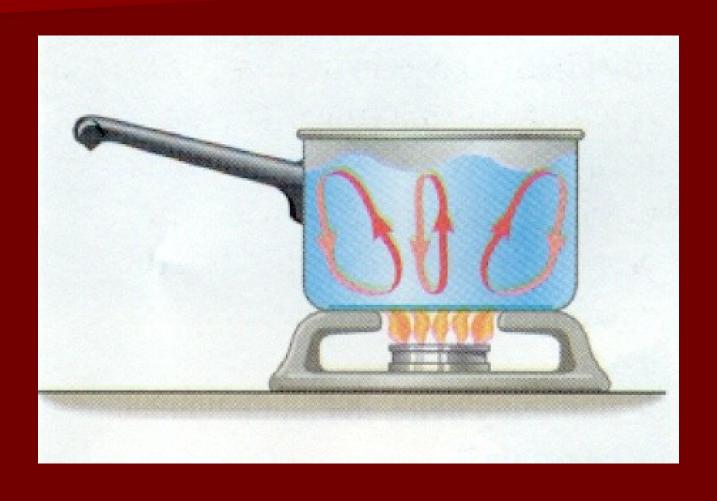
v. Conduction

- A. <u>Conduction</u> is the transfer of thermal energy through the direct contact of particles.
- B. Heat flows from the warmer object into the cooler object until they both reach the same temperature.(thermal equilibrium)

- C. The faster moving particles in the warmer object collide with the slower moving particles in the cooler object.
 - 1. as they collide, they give up some of their energy to the slower moving particles.

- 2. the slower moving particles gain thermal energy and pass it along to other particles.
- D. Solids are better <u>conductors</u> than liquids, and liquids are better conductors than gases.
 - 1. metals are good conductors
 - 2. air is a bad conductor

- E. The ability to transfer heat within an object is called *thermal conductivity*.
 - 1. gold, silver and copper have high thermal conductivity.
 - 2. they are also good conductors of electricity.
 - 3. glass and wool have low thermal conductivity.
 - 4. they make good *insulators*.



VI. Convection

- A. <u>Convection</u> is the transfer of thermal energy through currents.
- B. Convection occurs in liquids and gases.
- C. As liquids and gases gain thermal energy, they expand and decrease in density.

- D. Warmer areas of liquids and gases rise to the cooler areas.
 - 1. the cooler areas then take the place of the warmer areas.
 - 2. this movement of particles creates a current of thermal energy throughout a substance.

